

# Molecular dynamics modeling of particle and molecule interaction with solid and liquid surfaces

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ALPS Meeting, Grand Canyon, AZ  
April 9-11, 2003

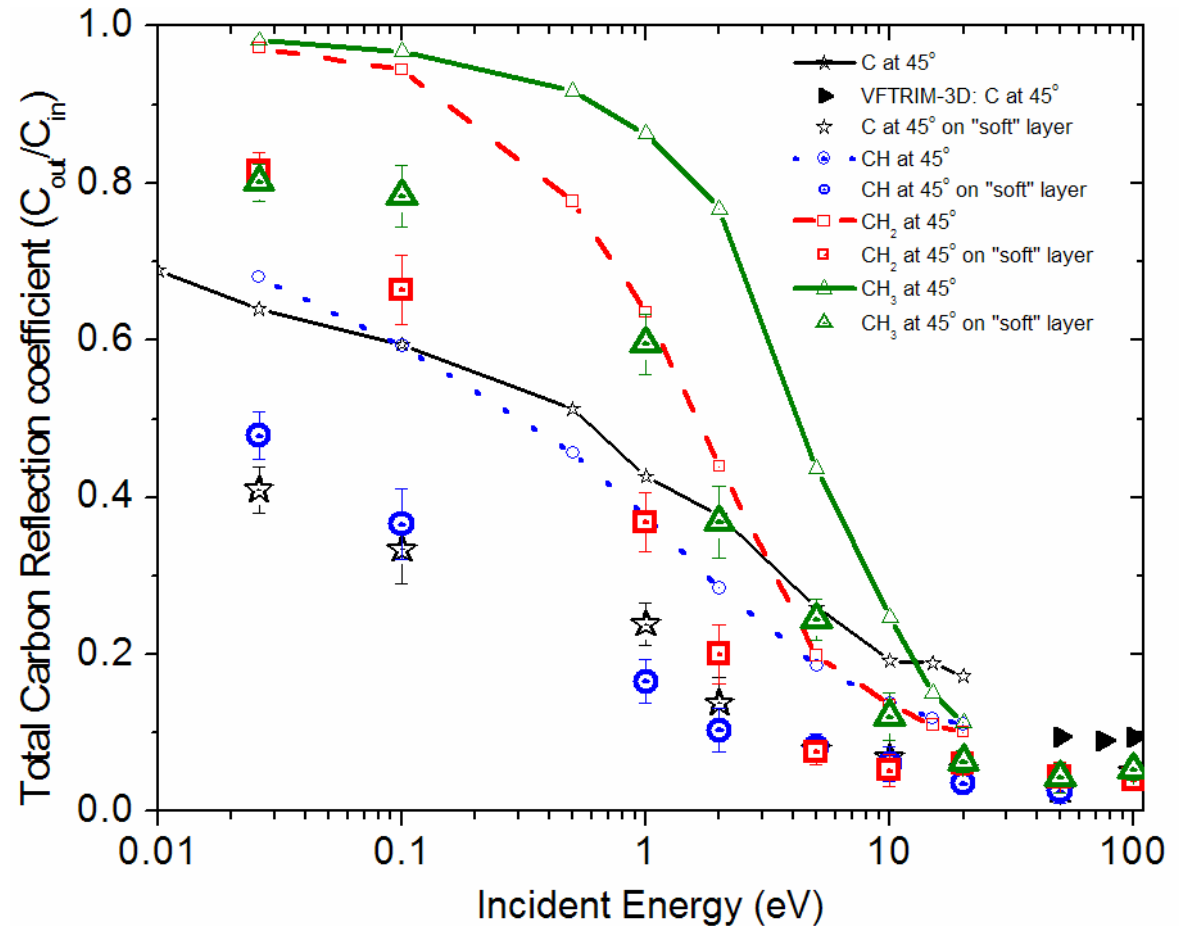


# Outline of Work at the UIUC

- Molecular Dynamics simulations of hydrocarbon plasma-material interaction
- Molecular Dynamics simulations of liquid lithium to study self-bombardment reflection
- Analytical studies of backscattered and sputtered charge fraction at low energies
- Liquid lithium erosion enhancement studies implementing a cascade model from molecular dynamics in a modified version of VFTRIM-3D
- Recent progress with FIRE first wall Be transport and divertor mixing issues

# MD Modeling of Hydrocarbons

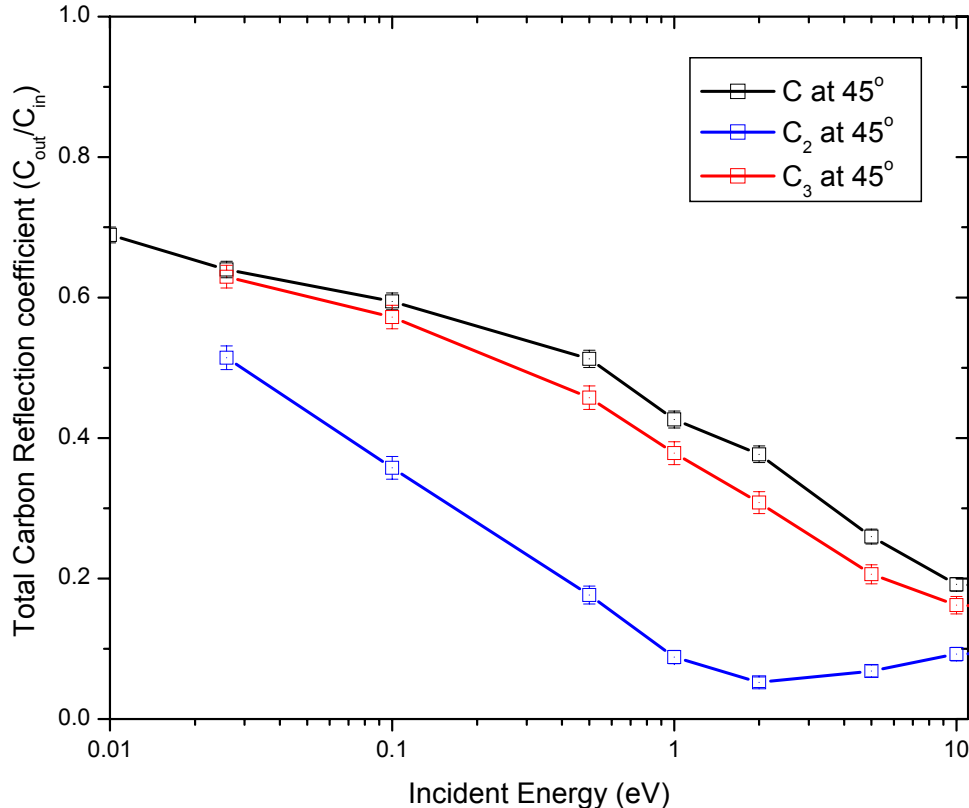
- First surface used was graphite bombarded by hydrogen until saturation ( $\sim 0.4$  H:C)
- Second surface - a "soft" H:C surface
  - Formed by deposition of thousands of hydrocarbons on a graphite surface
  - Emulates a redeposited carbon layer in a tokamak
    - Larger H:C ratio in this redeposited layer
    - Lower density
    - Carbon is less strongly bound
- Results show reflection is lower on the "soft" surface than the 0.4 H:C graphite at one energy (1 eV)
- Work continues to look at a full range of incident energies



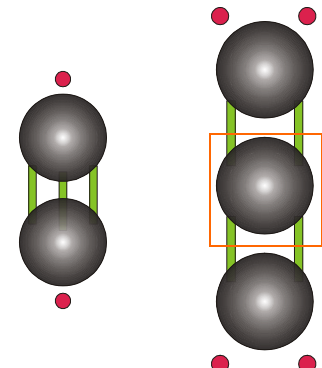
# Reflection from “soft” vs. “hard” surfaces

- Why is reflection lower on the “soft” surface
  - The top layer of the “soft” surface is less dense
  - Incident atoms/molecules can penetrate more easily
  - Once inside the surface, the particles are more likely to be trapped there
  - On the “hard” surface, incident particles are more apt to have a hard collision on the very top surface and bounce off
  - Observation of more movies may provide further insight

# Reflection of carbon dimer and trimer molecules



- Why does C<sub>2</sub> stick more readily than C<sub>3</sub>?
  - One reason could be the fully bonded atom at the center of the trimer molecule
  - The end atom may attempt to stick, but when the molecule swings into position a repulsive force is created

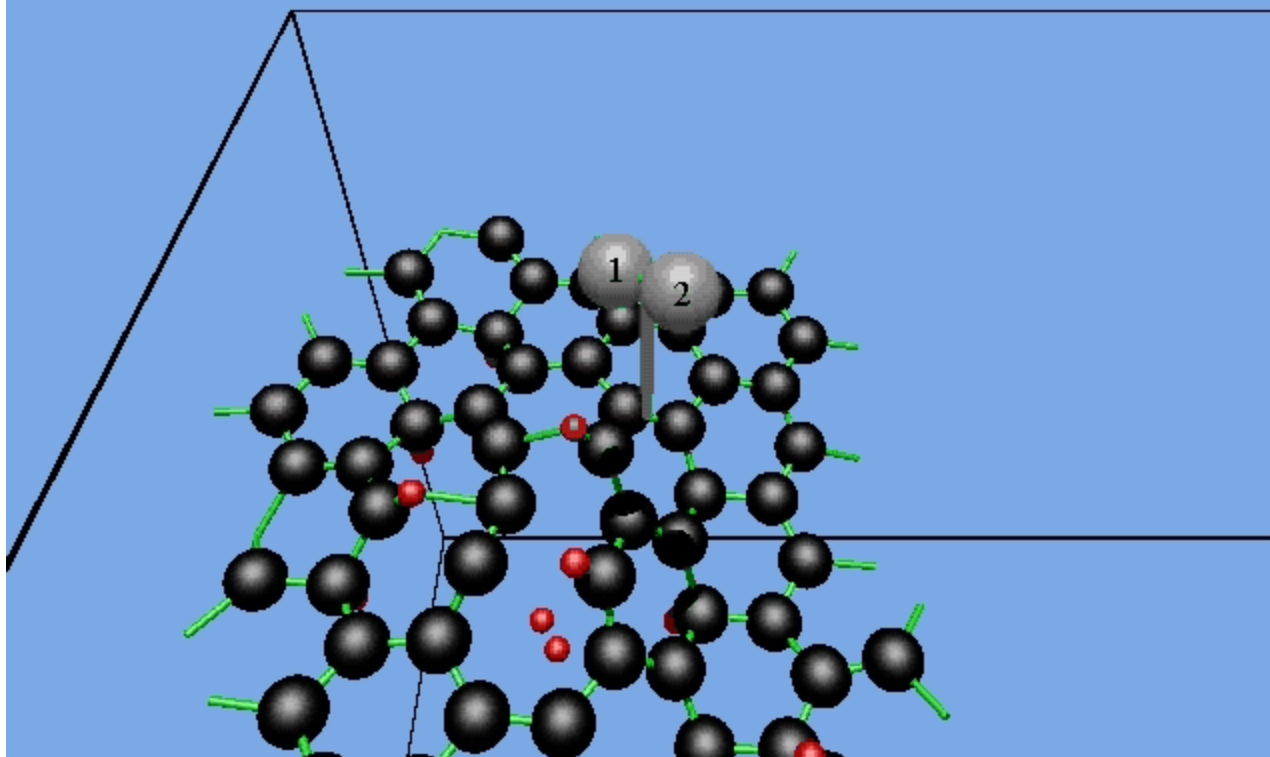


# Carbon dimer molecules tend to stick readily

214350336

C2 incident @ 2 eV, 45 degrees

Does not reflect

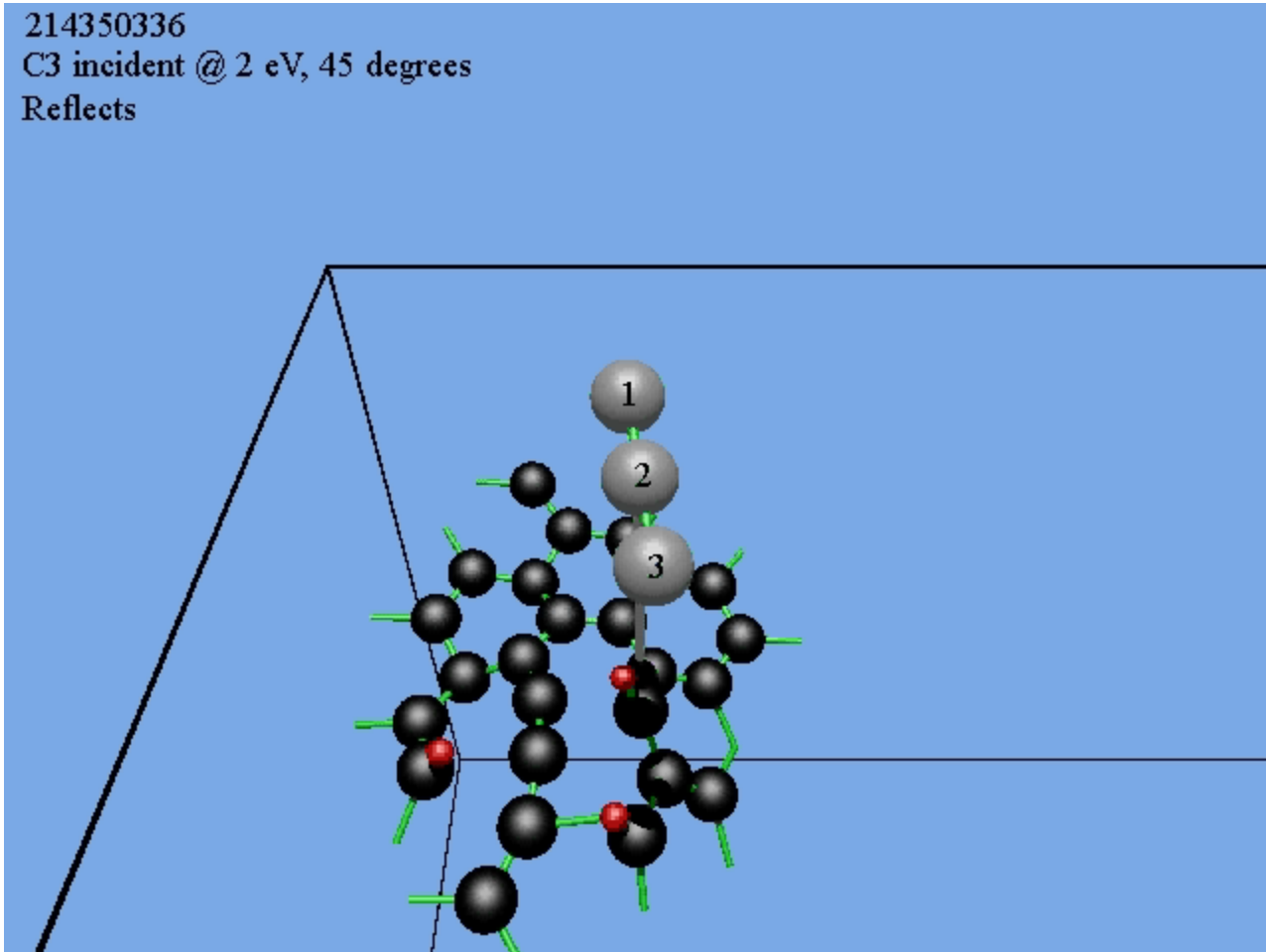


# Carbon trimers are more likely to bounce off

214350336

C3 incident @ 2 eV, 45 degrees

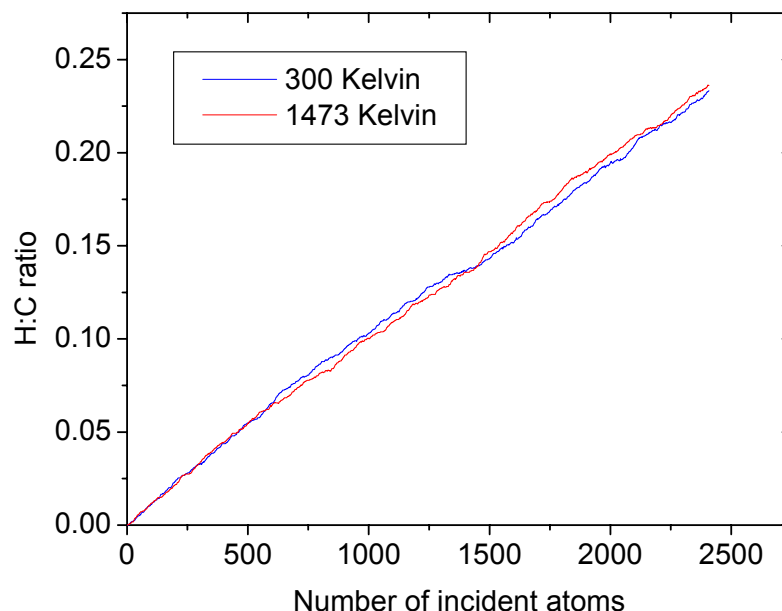
Reflects



# Multi-scale problems: long time scale issues

- For example, does the graphite surface saturate at a lower H:C ratio at higher temperature?
- First attempt showed no difference between hydrogen concentration during H bombardment at room temp. & 1200 °C
- Possible issues:
  - Timescale for desorption of hydrogen is long  $\sim 10^{-6}$ - $10^{-4}$  s
  - These MD simulations (e.g. to look at reflection) typically run for  $10^{-13}$ - $10^{-12}$  s
- Recent progress
  - Implemented a variable timestep algorithm in the MD code
  - Factor of  $\sim 3$  speedup vs. fixed timestep with comparable accuracy

## First attempt



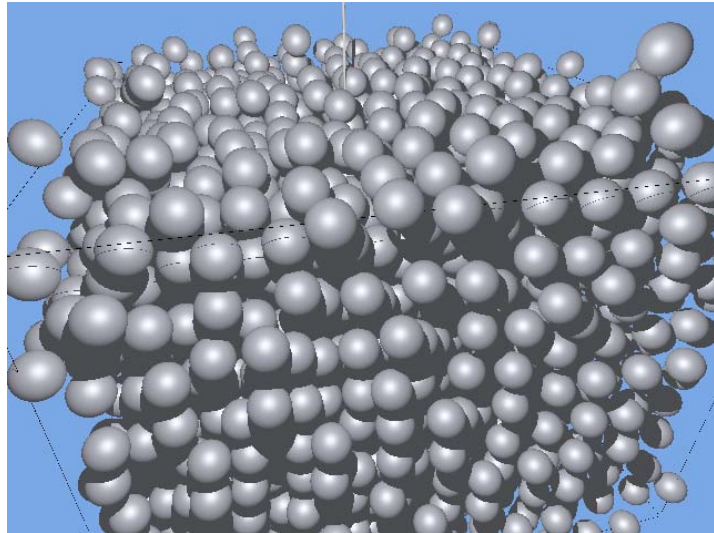
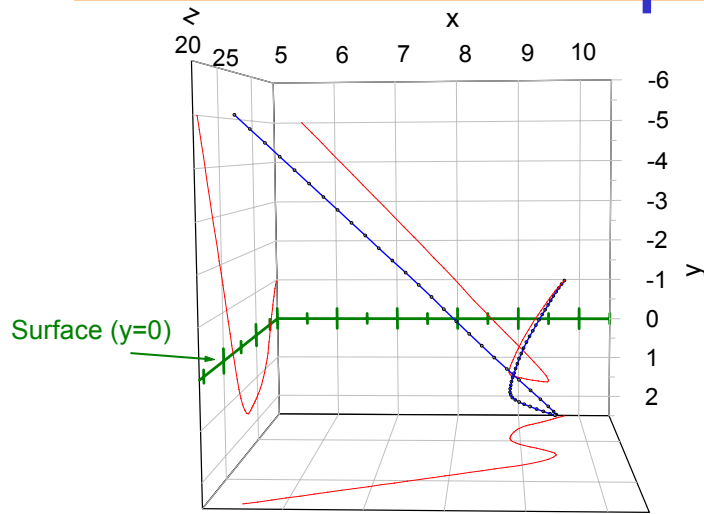
- Enabled a long (100 ps) run to look at H desorption from the H saturated graphite surface
- Hydrogen is visibly more mobile
- 41% more H evolved at higher T, but still not conclusive



# Modeling of reflection and charge state of backscattered particles

- Sputtered lithium particles in the ionized state leave surface at about 1-5 eV and quickly return to surface due to the magnetic sheath potential. Need to determine backscattering of incident lithium ions.
- Semi-analytical models exist to determine the charge state of backscattered particles from alkali metals such as lithium.

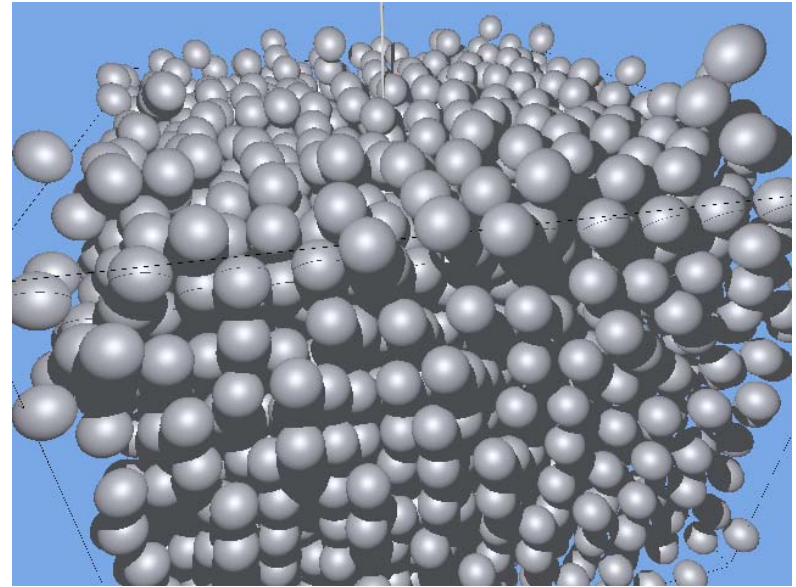
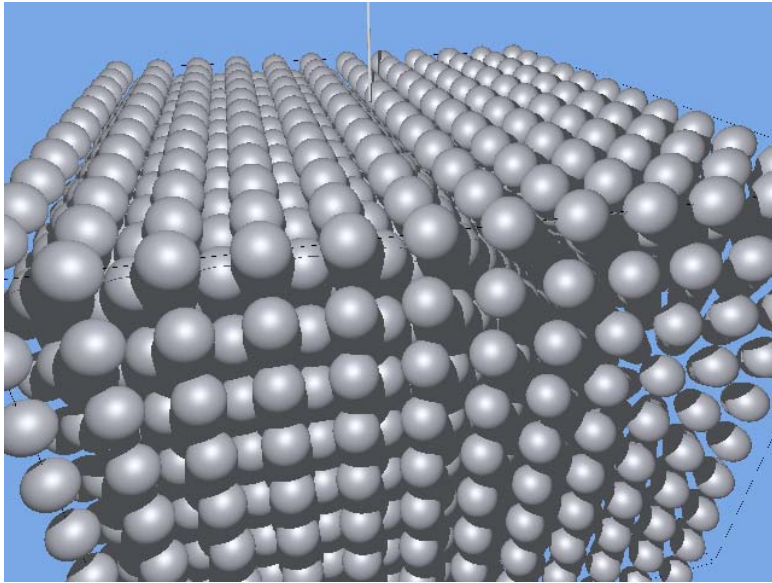
# MD modeling of lithium bombardment on liquid lithium surfaces



- Investigation of reflection of lithium atoms on liquid lithium surfaces continues for NSTX conditions
  - 0.35 and 2 eV incident energy
  - 20 degrees incident angle
  - 473 K and 653 K surface temperatures
- Major changes have been made to the code to better incorporate lithium
  - Enabling lithium runs to be integrated into the distributed computing system already in use for hydrocarbon modeling (giving ~10x speed-up)
  - Calculation of ion fraction of reflected/sputtered atoms now built in
  - New liquid lithium potential data included<sup>†</sup>

L.E. Gonzalez, private comm. 2002  
Universidad de Valladolid

# Liquid lithium simulation setup

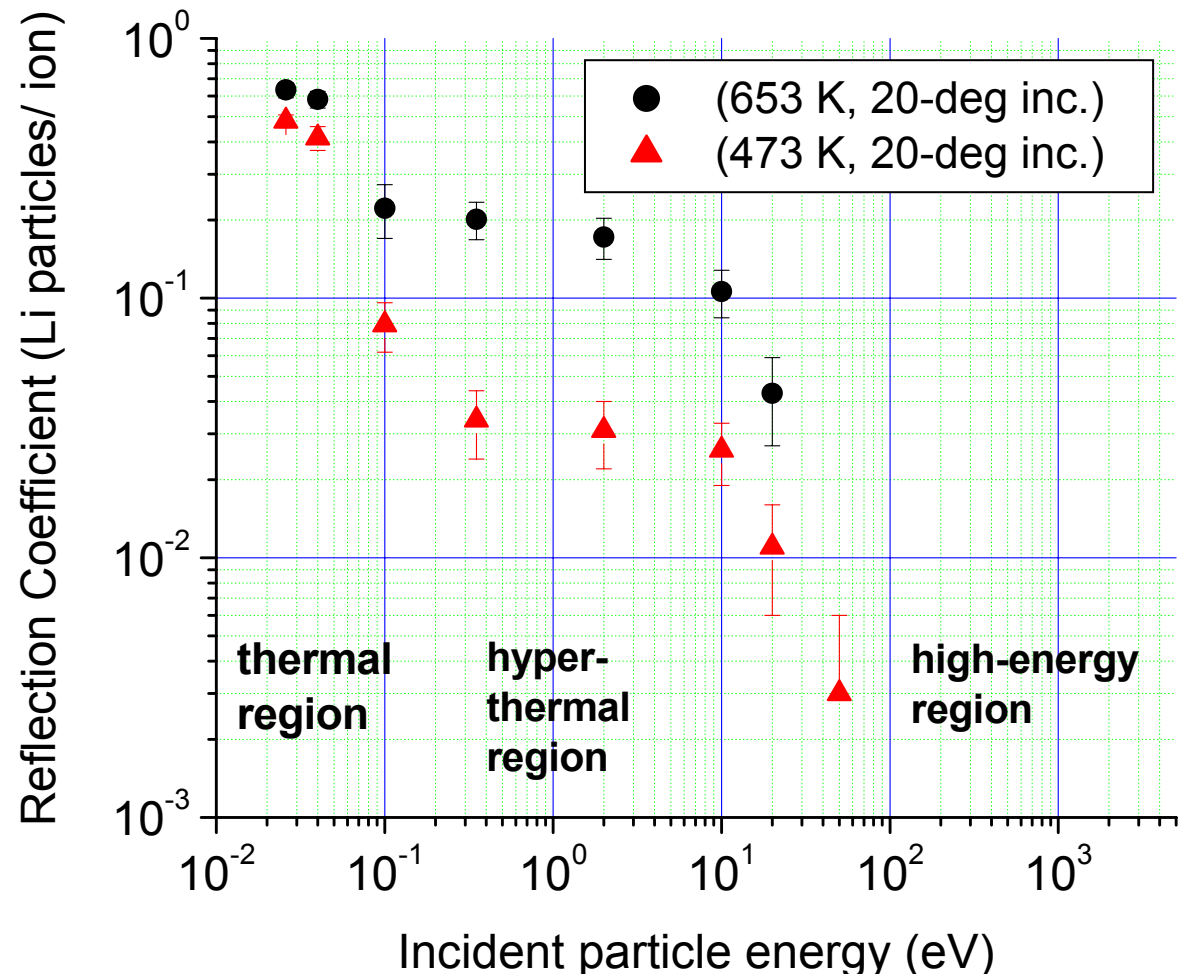


- Temperature control is achieved by using a simple velocity scaling technique at each time step<sup>1-3</sup> to maintain the desired temperature at the edges of the surface.
- The resulting target surface is an amorphous liquid lithium surface 42.2 by 42.2 Å and 34.2 Å deep.

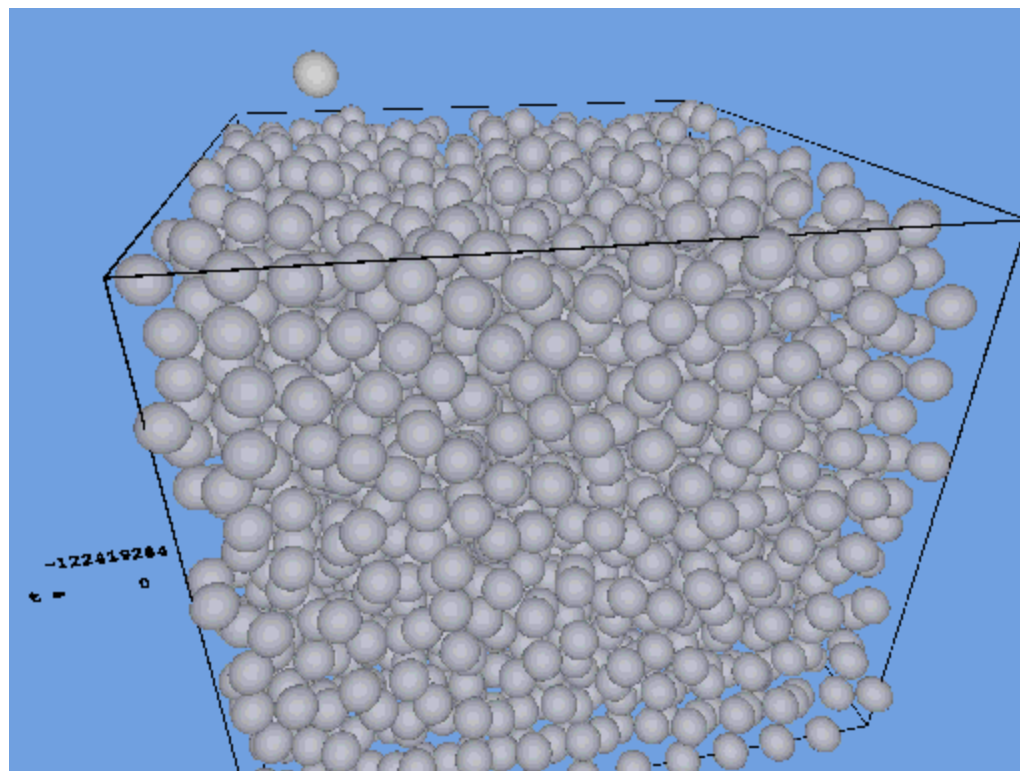
1. L. V. Woodcock, Chem. Phys. Lett. **10**, 257 (1970).
2. D. J. Evans, Mol. Phys. **37**, 1745 (1979).
3. T. Schneider and E. Stoll, Phys. Rev. B **13**, 1216 (1976).

# Molecular dynamics simulations of liquid lithium reflection

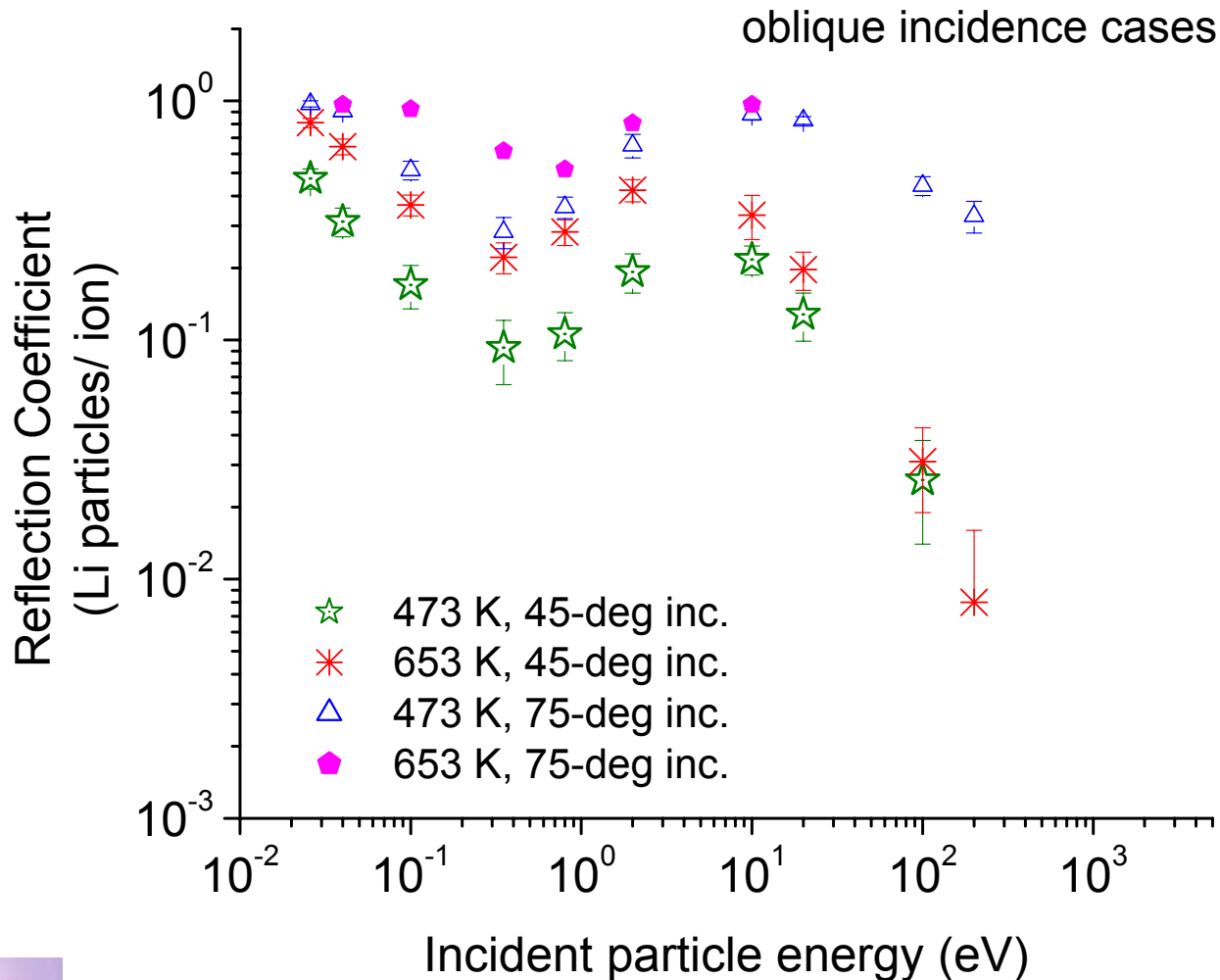
- NSTX cases: 473 and 653 K, 20-deg. incidence
- Reflection results show three distinct regions for low-energy self-bombardment reflection of lithium
- A region is found where the reflection coefficient varies little with incident energy (hyperthermal region)
- MD modeling continues to investigate this behavior as well as oblique bombardment (45,75-degree inc.)
- Other issues include: other temperatures and hydrogen treatment of lithium surface



0.35 eV, 20-deg inc, 653 K

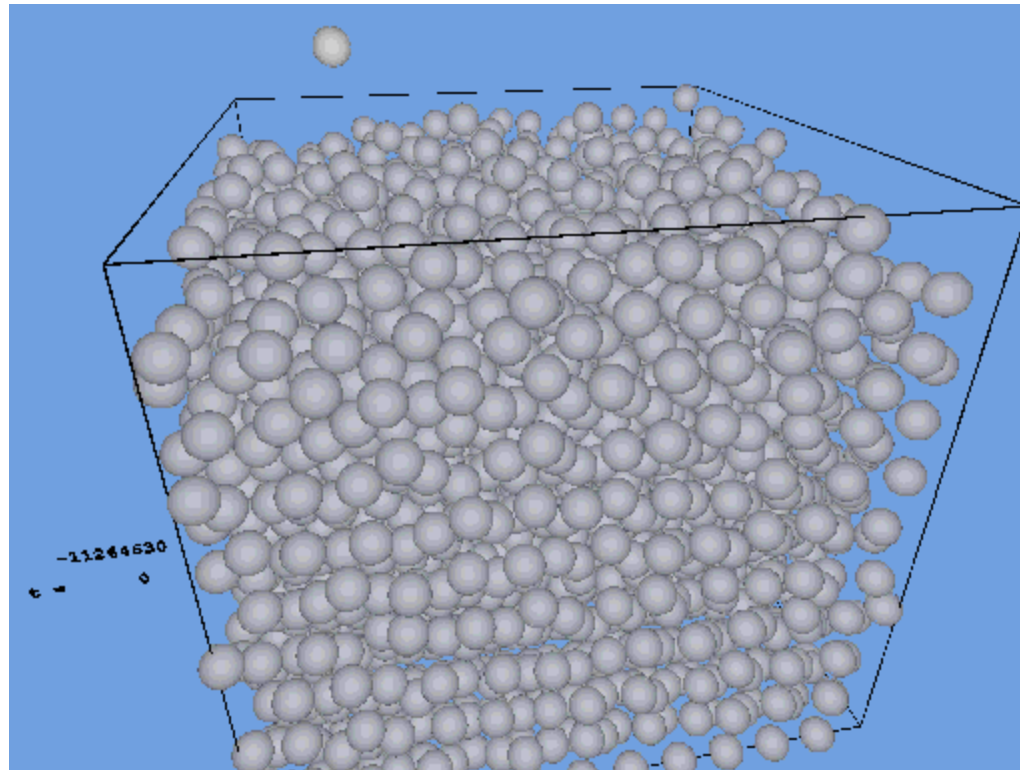


# Lithium reflection at oblique incidence

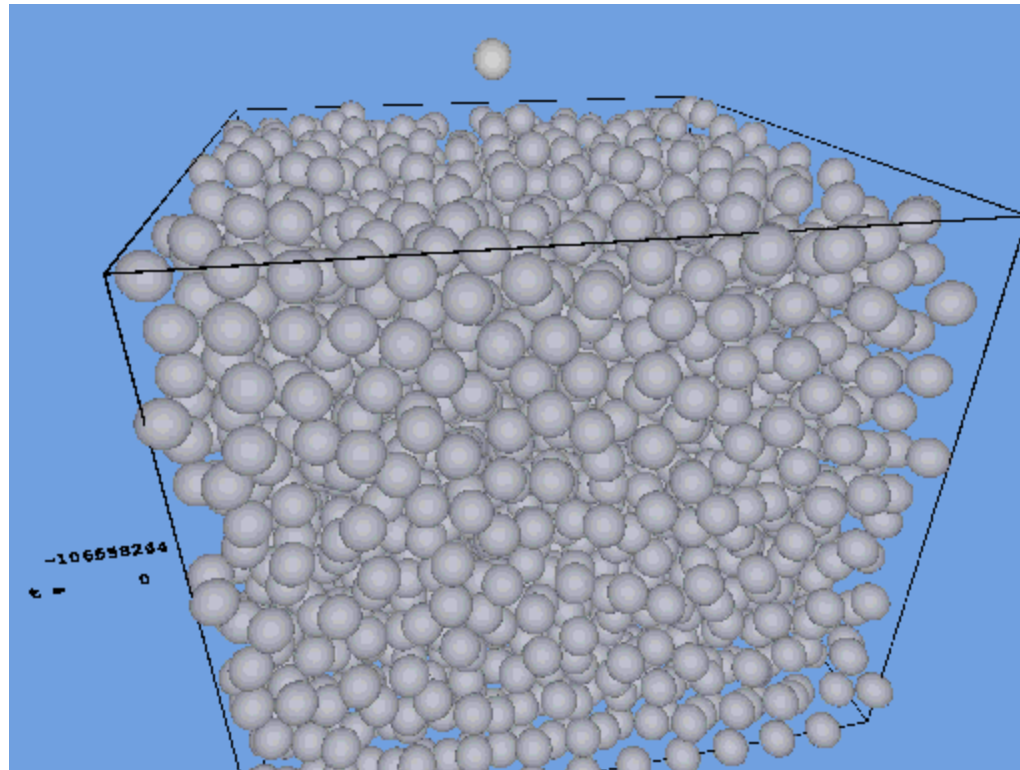




2.0 eV, 473 K at 45-deg inc.



2.0 eV, 653 K at 45-deg inc.





# MD/TRIM liquid Li erosion enhancement modeling studies

- Lithium erosion enhancement is studied using molecular dynamics of Li bombardment of liquid Li<sup>†</sup>
- Near-surface energy cascades are found from MD results for a variety of system temperatures
- The recoil energy and angular distributions are implemented in a Monte Carlo code (modified VFTRIM-3D) to obtain absolute lithium sputtering yields for comparison with experimental Li erosion data
- In addition, the surface binding energy from MD is implemented in VFTRIM-3D as a function of system temperature

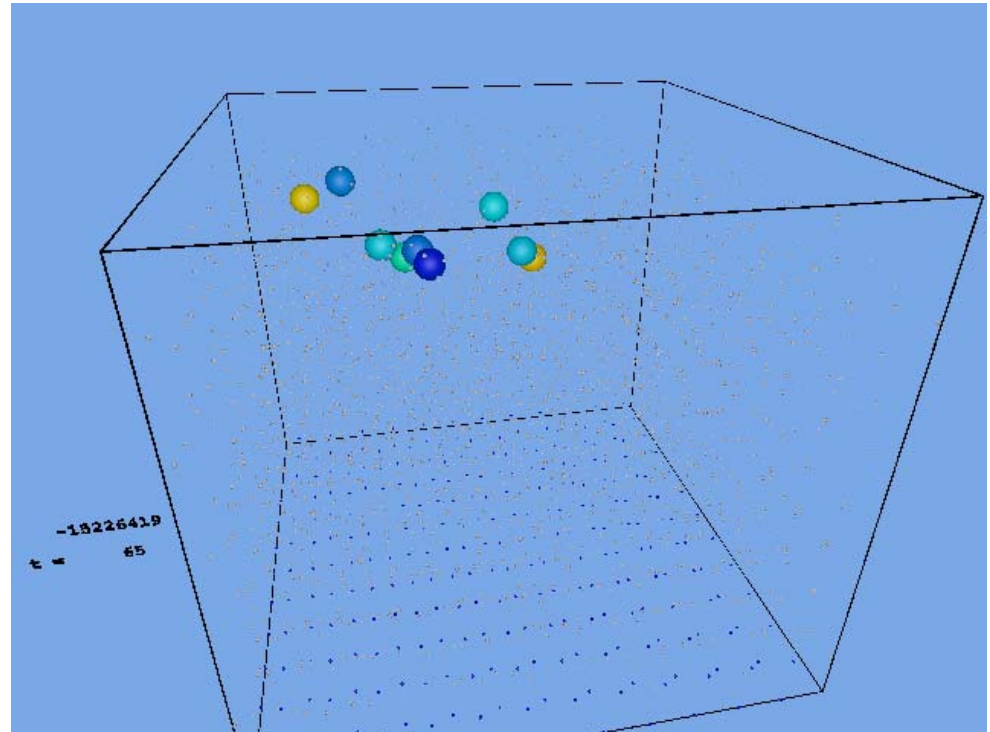
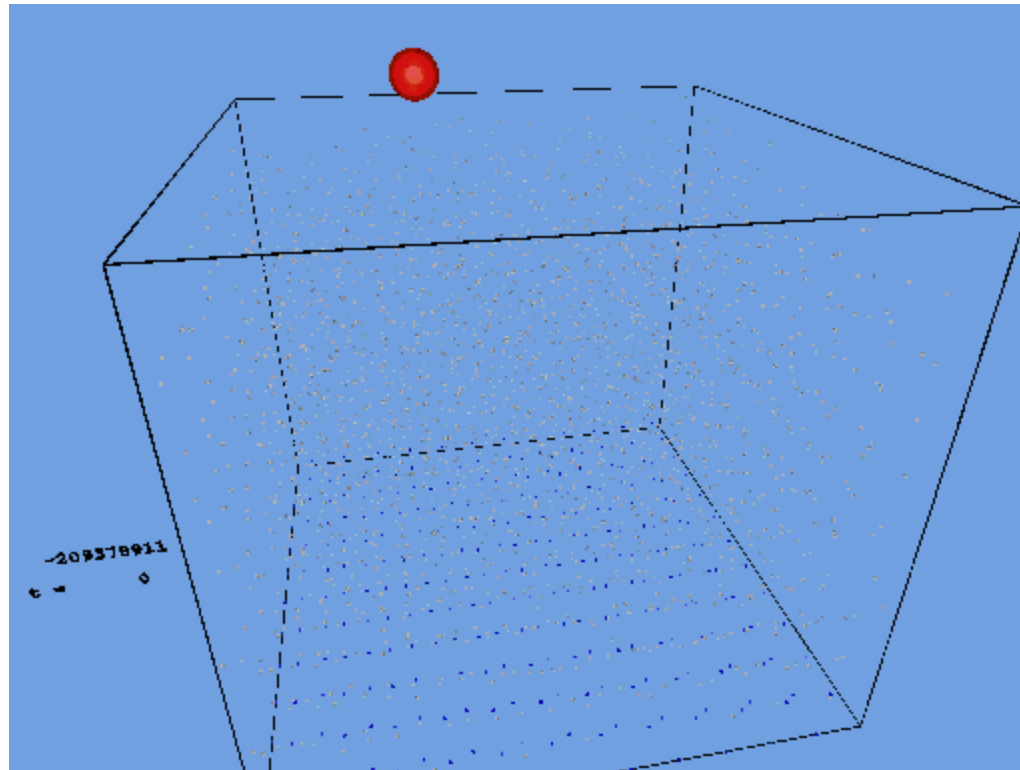


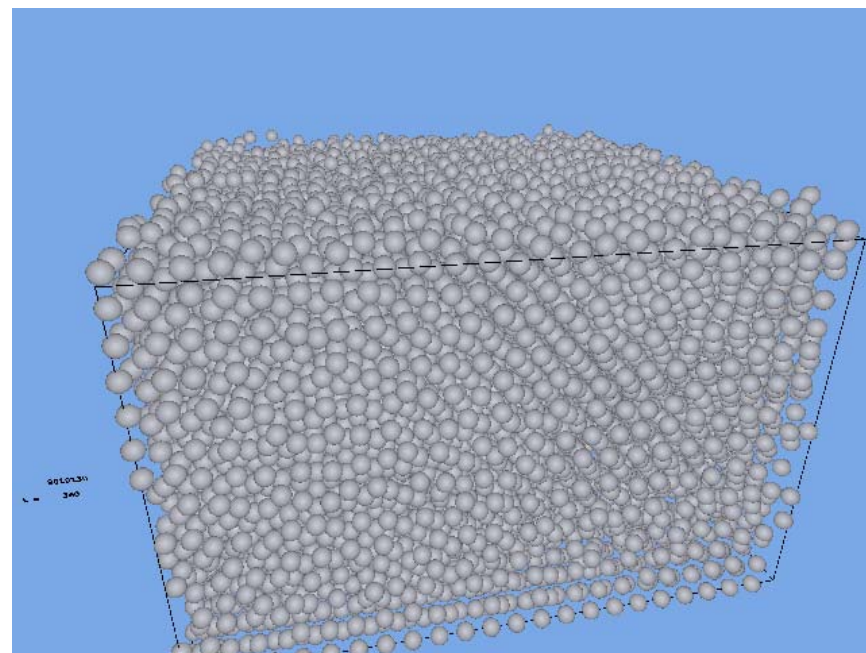
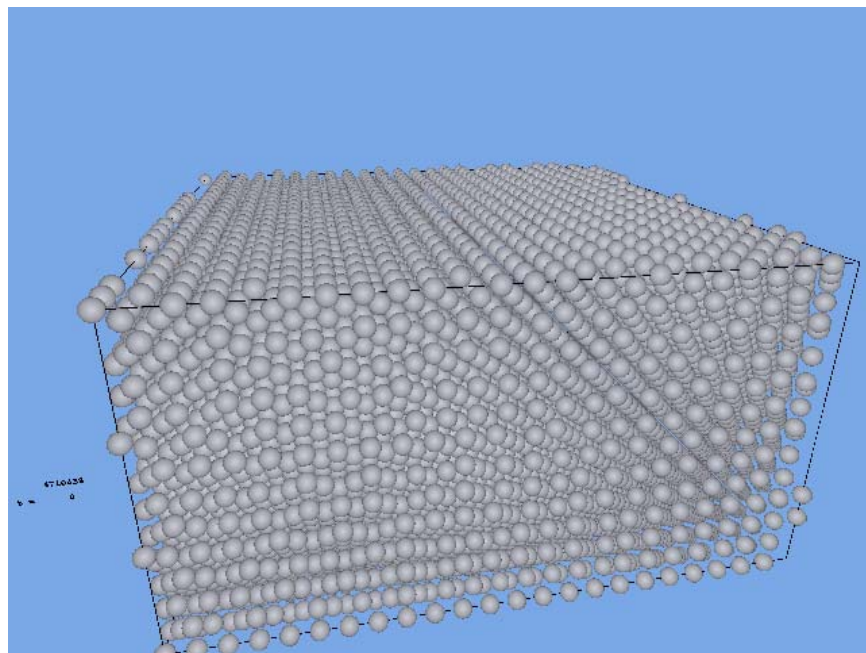
Figure shows near-surface energy cascade of recoils from a 100 eV, 45-deg. incident Li atom. Colors represent energy magnitudes (red – highest, blue - lowest). Only atoms with  $10 \cdot kT$  or more are shown.

<sup>†</sup> L.E. Gonzalez, private communication, 2002  
L.E. Gonzalez, J. Phys. Conds. Matter 5 (1993) 4283.

# 200 eV, 45-deg. Li on liquid Li, reflection and sputtering

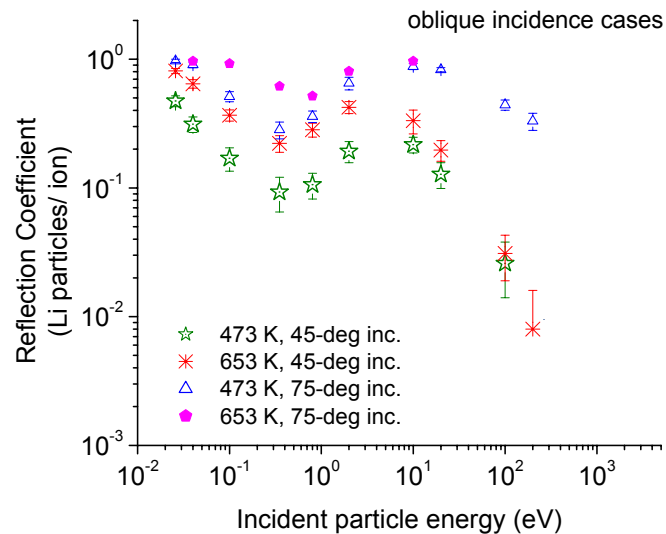


# Analysis of molecular dynamics simulations of liquid surfaces

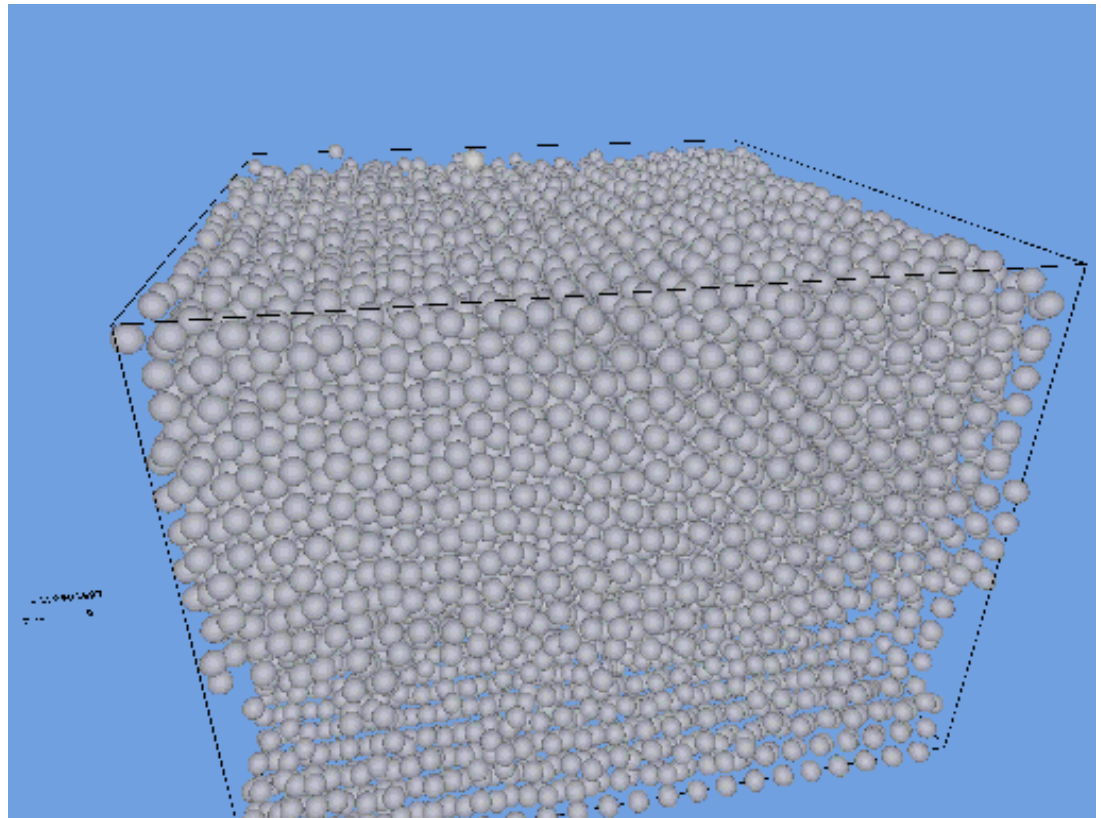


- Efforts begun in building larger lattices being mindful of computational expense.
- The size of the lattice may be relevant in modeling incident energies ranging from 300-700 eV. The effect of size on cascade dynamics are currently under investigation.
- A lithium lattice of about 13,300 lithium atoms is now in use and has been heated to temperatures of 473 and 653 K. Equilibration times ~ 25-250 picoseconds.

# Case for reflection with large lattice: 75-deg inc. 10 eV



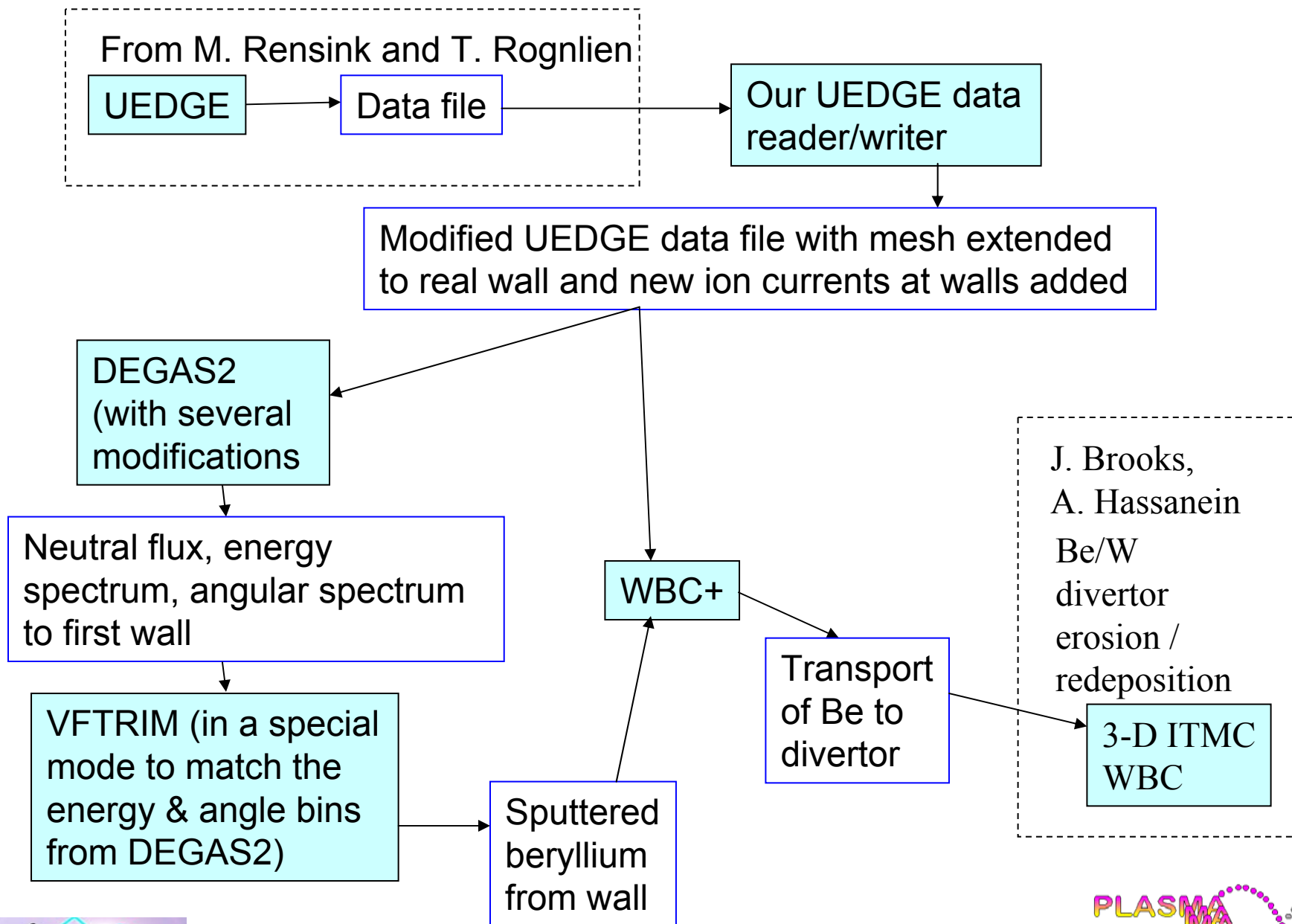
$T = 473 \text{ K}$



# NSO/FIRE Modeling

- [illegible]





# Conclusions and Future Work

- Work ongoing on reflection from “soft” and “hard” carbon surfaces
- $C_3$  reflects more than  $C_2$  due to bonding nature of carbon trimer
- Reflection for low-energy Li on liquid Li shows large yields at thermal and hyperthermal energies
- The yield rises with angle of incidence, however functional behavior is different
- More work continues coupling MD simulations with TRIM in understanding nature of enhanced erosion of liquid Li (in particular) or liquid-metals (in general)...more in tomorrow's talks: Allain and Ruzic
- Li particles that reflect at thermal or hyperthermal energies will mostly consist of neutrals
- Be currents of the order of  $10^{20} \text{ s}^{-1}$  reach the inner and outer divertor of FIRE